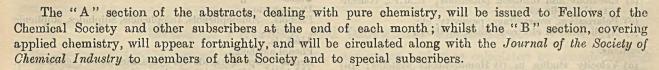
CHEMICAL ABSTRACTS

Foreword.



The price of the "A" and the "B" sections is £3 13s. 6d. each per annum, post free (including joint Index), but Fellows of the Chemical Society may obtain the "B" abstracts for £1 10s. 0d., whilst Members of the Society of Chemical Industry may obtain the "A" abstracts for £2 0s. 0d. [The yearly membership subscriptions are £3 0s. 0d. in the case of the Chemical Society and £2 10s. 0d. in the case of the Society of Chemical Industry. Copies of "A" or "B" abstracts printed on one side of the paper, and suitable for filing purposes, may be obtained at reasonable charges

The general basis of classification adopted in the two sections is printed below. For the guidance of readers of "A" abstracts, it should be pointed out that abstracts of analytical papers may be found not only at the end of each section as tabulated below, but sometimes also, when the analytical method described has a very specialised object, in the body of the section, according to the material with which the analytical method deals. In such cases there will be a reference in the "analytical" portion to other abstracts of analytical interest appearing in the same section.

A.—PURE CHEMISTRY.

General, Physical, and Inorganic Chemistry.

Sub-atomics.

- (a) Atomic spectra. Infra-red, visible, ultra-violet, X-ray emission and absorption spectra, Zeeman and Stark effects, Compton effect.
- (b) Electrical properties: Ionisation potentials of atoms, photo-electric and thermionic effects.
- (c) Properties of electrons and gaseous ions. Magnetic properties.
- (d) Isotopes—atomic weights. Radioactive processes.
- Other sub-atomic processes.
- Theories of atomic structure and sub-atomic mechanism.
- (g) Theories of atomic structure and state).

 (h) Atomic dimensions (except in solid state).

Molecular Structure.

- (a) Molecular spectra: Emission and absorption spectra of organic and inorganic substances. Fluorescence, luminescence, and phosphorescence. Raman effect.
- (b) Ionisation potentials. Photo-electric effect with compounds.
- (c) Conductivity. Dielectric constants. Dipole moment.
- (d) Molecular volumes.
- (e) Optical properties: Molecular refraction, dispersion, dispersion, optical activity, magnetic rotatory rotation.
- (f) Theories of molecular structure. Valency, secondary valency, including co-ordination, electronic and magnetic theories, constitutional formulæ of inorganic substances.
- (g) Molecular sizes and forces. Surface tension.

Crystal Structure.

- (a) X-Ray examination.
- (b) Crystal models.

- (c) Magnetic and electrical properties of crystals: Piezoelectricity; magnetostriction.
- (d) Optical properties. Rotatory dispersion.
 (e) Compressibility. Tensile strength.
 (f) Mesomorphic state.

Physical Properties of Pure Substances (not included above).

- (a) Molecular weights.
- (b) Electrical constants: Conductance, thermoelectric power, light sensitivity, etc.
- Optical constants.
- (c) Optical constants.
 (d) Thermal constants: Specific heats, heat of change of state, boiling points, freezing points, transition points.
- (e) Chemical constants.
- Pressures and volumes: Density, vapour pressure, coefficient of expansion, equations of state, theory of corresponding states.
- Compressibility
- Viscosity; fluidity.
- (i) Diffusion.

Solutions and Mixtures.

- (a) Gaseous mixtures, liquid mixtures (excluding dilute solutions), solid solutions (including alloys), propertycomposition curves.
- (b) Miscibility of liquids and of solids. Solubility of gases and solids in liquids.
- (c) Distribution phenomena: Partition, absorption, adsorp-
- tion, surface films, surface energy, membrane effects.

 (d) Dilute solutions: (i) Non-electrolytic solutions; (ii) Solutions of electrolytes. Colligative properties;
- non-colligative properties.
 (c) Disperse systems. Preparation and properties of suspensions, emulsions, smokes, foams, sols, gels, jellies. Coagulation, peptisation, ageing, cataphoresis, imbibi-

Kinetic Theory. Thermodynamics.

- (a) Equilibrium in homogeneous systems; equilibrium, dissociation, ionisation constants, activity coefficients,
- (b) Equilibrium in heterogeneous systems; uni- and multicomponent systems, phase rule.
- (c) Thermochemistry.

Electrochemistry.

- (a) Electrical conductance.
- Transport phenomena.
- (c) Electrode and diffusion potentials; E.M.F., concentration cells, etc.
- (d) Polarisation, overvoltage, passivity, etc.
- (e) Application of electrochemical methods.

- (a) Velocity studies in (i) Homogeneous systems; (ii) Heterogeneous systems.
- (b) Catalysed reactions: (i) and (ii) as above.
- (c) Electrode reactions.
- (d) Photochemical reactions.
- (e) Irradiated reactions.

New Methods of Preparing Substances (arranged according to periodic table), etc.

Improved Methods of Preparing Substances, etc.

Analysis. Apparatus.

Lecture Experiments.

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Organic Chemistry.

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Hydrocarbons.

Halogen, nitro-, and nitroso-derivatives.

Alcohols.

Ethers.

Alkyl salts.

Sulphur compounds, including sulphonic acids. Acids.

Thio- and sulpho-acids.

Aldehydes.

Aldoximes.

Aldoximes.

Ketones and diketones.

Ketoximes.

Ketoximes.

Sugars, glucosides, and carbohydrates.

Amines.

Amino-alcohols.

Cyano-acids,
Amino-aldehydes and -ketones,
mides (including

Amides (including cyanic, cyanuric, and thiocyanic acids). Nitriles, carbylamines, metallic cyanides.

Amidoximes, imino-ethers.

Diazo compounds.

Phosphorus compounds.

Arsenic, antimony, boron, silicon, etc. compounds.

Aliphatic organo-metallic compounds.

Homocyclic.

Hydrocarbons C_nH_{2n} to C_nH_{2n-6} .

Halogen, nitroso-, and nitro-derivatives.

Sulphonic acids.

Hydrocarbons C_nH_{2n-8} to C_nH_{2n-4}. Derivatives under each.

Amines. Includes anilides, carbamides, thiocarbamides, and sulphonic acids.

Azoxy compounds.

Azo

Hydrazo Diazo

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Diazoamino "...
Phenols. Aminophenols, thiophenols, sulphides, sulphonic acids.

Alcohols.

Phenol-alcohols.

Acids. Sulphonic acids and hydrazines under each member. Aldehydes. Derivatives under each.

Camphor group. "Hanhard beamant) to zboice and to

Terpenes. Ethereal oils.

Resins and balsams.

Resins and balsams.

Bitter principles and indifferent substances.

Colouring matters (unclassified natural and artificial). Tannins.

Heterocyclic, etc.

Thiophen group (Se).

Bases: N₁, N₂, N₃, etc.

Alkaloids.

Aliphatic, and the land of the second of the Phosphorus compounds.

As, Sb, Bi, B, Si compounds.
Organo-metallic compounds.

Proteins.

Analysis.

Biochemistry.

Respiration.

Blood: Gases; constituents; reactions (hæmolysis, anti-body formation; etc.).

Organs and their Proximate Constituents: Analytical data of constituents of organs: pure substances isolated from organs. Secretions: Milk; lymph, etc.

Excretions: Urine; fæces; other excretions.

Diseases: Natural and experimental (in alphabetical order). Metabolism: General; special; intermediary; fate of substances in the animal body.

Physiological Action: Variation of physiological conditions; action of drugs; toxicology. properties

Enzymes: General; specific.

Micro-organisms: yeasts, moulds, protozoa, bacteria. Hormones.

Vitamins.

Vegetable Physiology: General; reproduction and fertilisation; respiration; growth; proximate principles; diseases; poisons. Analysis. margania bas not share a soboge subsceled (a)

B.—APPLIED CHEMISTRY. To the state of the st

- General; Plant; Machinery. Fuel; Gas; Tar; Mineral Oils.
- II. Organic Intermediates. III.
- IV. V. Dyestuffs.
- Fibres; Textiles; Cellulose; Paper.
- Bleaching; Dyeing; Printing; Finishing.
 Acids; Alkalis; Salts; Non-Metallic Elements.
 Glass; Ceramics. VII.
- VIII. Building Materials.
- IX. X. Metals; Metallurgy, including Electrometallurgy.
- Electrotechnics. XII. Fats; Oils; Waxes.
- Paints; Pigments; Varnishes; Resins. Indiarubber. Leather and Glue.
- XIV. XV.
- XVI. Agriculture.
- XVII.
- rotatory dispersion, Agriculture.
 Sugars; Starches; Gums.
 Fermentation Industries. XVIII.
 - XIX. Foods.
 - XX.
- Medicinal Substances; Essential Oils. Photographic Materials and Processes. XXI.
- XXII. Explosives.
- Sanitation; Water Purification. XXIII.