BRITISH CHEMICAL AND PHYSIOLOGICAL ABSTRACTS

tion, surface films, surface energy, manbrane energy.

(d) Dilute solutions: (i) Non-electrolytic solutions; (ii)

Solutions of electrolytics. (eligative properties.)

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Topic configurative properties.

Topic configurative properties. Section "A" of the abstracts, dealing with pure chemistry and physiology, will be issued to Fellows of the Chemical Society and other subscribers at the end of each month, in three separate sections: A., I, General, Physical, and Inorganic Chemistry and Geochemistry; A., II, Organic Chemistry; A., III, Physiology and Biochemistry. Each section will be paginated separately and have its own monthly author index.

Section "B," covering applied chemistry, will appear at the end of each month, and will be circulated with the Transactions of the Society of Chemical Industry to members of that Society and to special subscribers.

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The general basis of classification adopted in the various sections is printed below. For the guidance of readers of "A" abstracts, it should be pointed out that abstracts of analytical papers may be found not only at the end of each section as tabulated below, but sometimes also, when the analytical method described has a very specialised object, in the body of the section, according to the material with which the analytical method deals.

In order to facilitate reference to the "A" abstracts, the appropriate number and letter (1 c, 1 v d, etc.), as shown in the accompanying statement, will be printed at the top of each page on which the respective abstracts appear.

Sees boold A.-PURE CHEMISTRY AND PHYSIOLOGY. Intel Comment of the State of the Stat

I. General, Physical, and Inorganic Chemistry.

I. Sub-atomics.

(a) Atomic spectra. Infra-red, visible, ultra-violet, X-ray emission and absorption spectra, Zeeman and Stark effects. Compton effect.

(b) Electrical properties: Ionisation potentials of atoms, photo-electric and thermionic effects.

- (c) Properties of electrons, positrons, and gaseous ions.

 Magnetic properties. Ramsauer effect.
- (d) Isotopes—atomic weights.
- Radioactive processes. Neutrons. da laistanda
- Other sub-atomic processes. Artificial radioactivity. Nuclear chemistry. Cosmic rays. as Issis
- Theories of atomic structure and sub-atomic mechanism.
- (h) Atomic dimensions (except in solid state).

II. Molecular Structure.

- (a) Molecular spectra: Emission and absorption spectra of organic and inorganic substances. Fluorescence, luminescence, and phosphorescence. Raman effect.
- pounds. Rectifiers.
 Conductivity. Dielectric constants. Dipole moment. Electrets.
- (d) Molecular volumes.
- (e) Optical properties: Molecular refraction, dispersion, rotatory dispersion, optical activity, magnetic rotation. Kerr effect.

 (f) Theories of molecular structure. Valency, secondary
- valency, including co-ordination, electronic and magnetic theories, constitutional formulæ of inorganic substances. Deuterium and its compounds.
- (g) Molecular sizes and forces. Surface tension. Molecular beam. Parachor.

III. Crystal Structure.

- (a) X-Ray examination of ried bas spod morabell (a)
- (b) Crystal models. X-Ray results for elements and compounds. Electron diffraction.
- (c) Magnetic and electrical properties of crystals: piezo-electricity; magnetostriction; Hall effect; Barkhausen effect; tribo-electricity.

 (d) Optical properties. Rotatory dispersion.

 (e) Compressibility. Tensile strength. Plasticity.

 (f) Mesomorphic state; allotropy; monotropy.

IV. Physical Properties of Pure Substances (not included above).

- (a) Molecular weights.
- (b) Electrical constants: Conductance, superconductivity, thermoelectric power, light-sensitivity, etc. Magnetic susceptibility. Sonic properties. ogmos osat
- (c) Optical constants.
 - (d) Thermal constants: Specific heats, heat of change of state, boiling points, freezing points, transition points.
 - (e) Chemical constants.
- (f) Pressures and volumes: Density, vapour pressure, co-(b) Ionisation potentials. Photo-electric effect with compounds. Rectifiers.

 Photo-electric effect with compounds. Rectifiers.

 Corresponding states, critical state. Heat-conductive and the conductive states are conductive states. Thermal accommodation. Joule-Thomson effect.
 - (g) Compressibility.
 - (h) Viscosity; fluidity; plasticity.
 - (i) Diffusion.

v. Solutions and Mixtures.

- (a) Gaseous mixtures, liquid mixtures (excluding dilute solutions), solid solutions (including alloys), propertycomposition curves.
- (b) Miscibility of liquids and of solids. Solubility of gases

(c) Distribution phenomena: Partition, absorption, adsorption, surface films, surface energy, membrane effects.

(d) Dilute solutions: (i) Non-electrolytic solutions; (ii) Solutions of electrolytes. Colligative properties; non-colligative properties.

(e) Disperse systems. Preparation and properties of suspensions, emulsions, smokes, foams, sols, gels, jellies. Coagulation, peptisation, ageing, cataphoresis, imbibition, etclarages against at althour does to bu

VI. Kinetic Theory. Thermodynamics.

- (a) Equilibrium in homogeneous systems; equilibrium, dissociation, ionisation constants, activity coefficients,
- (b) Equilibrium in heterogeneous systems; uni- and multicomponent systems; phase rule.

(c) Thermochemistry.

VII. Electrochemistry.

(a) Electrical conductance. 10 zelout initio gailerlear anum

Transport phenomena.

(c) Electrode and diffusion potentials; e.m.f., concentration cells, etc.

Polarisation, overvoltage, passivity, etc.

(e) Application of electrochemical methods.

vin. Reactions. Its vision Issuered out to orga order

(a) Velocity studies in (i) Homogeneous systems; (ii) Heterogeneous systems.

Catalysed reactions: (i) and (ii) as above.

Electrode reactions.

(e) Irradiated reactions. The property land to the land a trade IX. New or Improved Methods of Preparing Substances (arranged according to periodic table) etc.

x. Analysis.

XI. Apparatus.

XII. Lecture Experiments and Historical. Linearing a of the XIX.

XIII. Geochemistry.

II. Organic Chemistry. Old YHO MA YH

XIV. Aliphatic.

(a) Hydrocarbons and their halogen, nitro-, and nitrosoderivatives.

(b) Alcohols. Ethers. Alkyl salts. Sulphur compounds, including sulphonic acids.

(d) Aldehydes. Aldoximes.

(e) Ketones and diketones. Ketoximes org landq() (b)

Sugars, glucosides, and carbohydrates.

Amines. Amino-alcohols. Amino-acids, Cyano-acids, thiocyano-acids. Amino-aldehydes and -ketones.

(h) Amides (including eyanic, eyanuric, and thiocyanic acids).

(i) Nitriles, carbylamines, metallic cyanides.

(j) Amidoximes, imino-ethers. woquad paleominds

(k) Diazo-compounds.

(l) Phosphorus, arsenic, antimony, boron, silicon, etc. compounds.

(m) Aliphatic organo-metallic compounds of other

xv. Homocyclic.

(a) Hydrocarbons C_nH_{2n} to C_nH_{2n-6} , and their halogen, nitroso-, and nitro-derivatives. Sulphonic acids.

(b) Hydrocarbons C_nH_{2n-8} to C_nH_{2n-4} and derivatives.
 (c) Amines. Includes anilides of aliphatic acids, carb-

amides, carbamates, and sulphonic acids. Diamines and polyamines.

(d) Azoxy-compounds. (e) Azo 1708dra, tello bus veleca isomesia eli lo

General, Physical and Inorganic Cushisozard, (t) (g) Diazo-

Diazoamino-li, notinas dona vytamodonic (i) Phenols and their substitution products; phenol ethers; aminophenols; thiophenols; sulphides; sulphonic

(j) Alcohols; amino-alcohols; di- and tri-arylearbinols and their derivatives; sterols.

(k) Carboxylic acids; includes nitriles of the respective acids, amides, and hydrazides; substitution products.

Aldehydes and their derivatives.

(m) Ketones

(n) Quinones; benzoquinone, naphthaquinone, anthraquinone, and others; their substitution derivatives.

(o) Terpenes.

XVI. Miscellaneous unclassifiable substances.

XVII. Heterocyclic etc.

(a) Oxygen ring compounds.
(b) Sulphur ring compounds.

Compounds containing both O and S rings.

(d) Photochemical reactions. Dollars at another and (d) Nitrogen ring compounds: one N.

more than one N.

Rings containing O and N, S and N, etc. (g) Alkaloids.

Organo-metallic compounds.
Proteins.

(i) Proteins.

XVIII. Analysis.

III. Physiology and Biochemistry.

nitided at the top of each responded which the respective

(a) Histology. Totale nativacamoons on an arrowal.

Blood and lymph.

Vascular system.

Respiration and blood gases. (d)

(e) Muscle.

Nervous system. conl bas decised a december .

Special senses.

Ductless glands, excluding gonads.

Reproduction, heredity, and experimental embryology. Digestive system. sqs nortgroads has notashire Liver and bile.

(k)(1)

(m) Kidney and urine.

(n) Other organs and tissues; body-fluids. Tumours.

Nutrition and vitamins. (0)

Metabolism, general and special. (p) Pharmacology and toxicology. (q)

Industrial physiology and hygiene. Radiations.

(8)

Physical and colloidal chemistry.

(u) Enzymes.

(v) Microbiological and immunological chemistry.

(w) Plant physiology.

(x) Plant constituents.

10 (y) Apparatus and analytical methods. The representation (a)

IV. Dyestuffs.

V.

VI. Bleaching; Dyeing; Printing; Finishing.

VII. VII. Acids; Alkalis; Salts; Non-Metallic Elements.

III. Glass; Ceramics.

IX. Building Materials.

VIII.

X. Metals; Metallurgy, including Electrometallurgy.
XI. Electrotechnics.
XII. Fats; Oils; Waxes.

atnalence instruction (s)

of the state in the property of the XIII. Plastics; Resins; Paints; Coating Compositions.

XV. Leather; Glue. semulov naluosloit (v) XVI.b Agriculture. naluosloit i settiogora laoita (s)

Sugars; Starches; Gums. Grain Violator Fermentation Industries. XVII.

XVIII.

XIX. Foods.

XX. Medicinal Substances; Essential Oils.

XXI. XXI. Photographic Materials and Processes.
XXII. Explosives; Matches.

XXIII. Sanitation; Water Purification. cular beam. Percebor.